

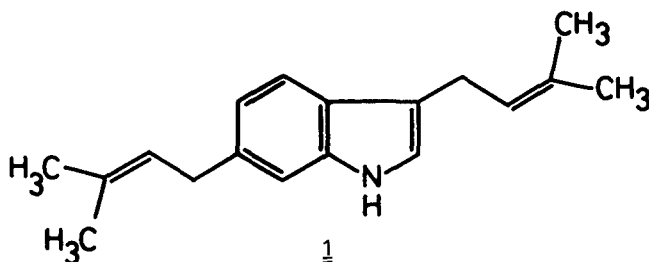
3,6-BIS(γ,γ -DIMETHYLALLYL)-INDOLE FROM UVARIA ELLIOTIANA

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Summary - The new title compound 1 has been isolated from the stem bark of Uvaria elliotiana Engl. & Diels (Annonaceae).

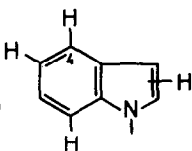
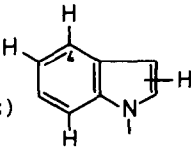
Chromatography (on SiO_2 with CHCl_3) of the crude methanol extract from the stem bark of Uvaria elliotiana, collected in West Africa, yielded as the main product a new alkaloid (0.15% of dry bark), for which structure 1 was deduced.



1 - $\text{C}_{18}\text{H}_{23}\text{N}$, $[\alpha]_{\text{D}}^{20} = \pm 0^\circ$, m.p. $< 20^\circ\text{C}$ - has an electronic spectrum [λ_{max} (MeOH) 293 (lg ϵ 3.72), 283 (3.89), 229 (4.65) nm] typical for an indole chromophore and an absorption band at 3480 cm^{-1} in the infrared spectrum (CHCl_3), which is characteristic for an $>\text{NH}$ -group.

Hydrogenation of 1 using Adams catalyst produces a liquid tetrahydro-derivative with an unchanged electronic spectrum.

The ^1H -nmr spectra of 1 and tetrahydro-1 show that the indole nucleus is disubstituted; the ^1H -nmr data consist of 5 groups of signals, which can be attributed to the structural features indicated in Table 1:

<u>1</u>		tetrahydro- <u>1</u>	
signal[ppm]	structural feature	signal[ppm]	structural feature
7.70 (1H, s broad, exchangeable)	>NH	7.68 (1H, s broad, exchangeable)	>NH
7.48 (1H, d, J=8Hz)		7.50 (1H, d, J=8Hz)	
7.10 (1H, s)		7.10 (1H, s)	
6.93 (1H, dd, J _O =8Hz, J _m =1Hz)		6.93 (1H, dd, J _O =8Hz, J _m =1Hz)	
6.83 (1H, d, J=1Hz)		6.83 (1H, d, J=1Hz)	
5.40 (2H, m)	2x -CH ₂ -CH=C<		
3.41 (4H, 2xd, J=7Hz)	2x >C-CH ₂ -CH<	2.70 (4H, 2xt, J=7Hz)	2x >C-CH ₂ -CH ₂ -
		1.57 (6H, m)	2x -CH ₂ -CH ₂ -CH<
1.72 (12H, weakly broadened)	4 x =C-CH ₃	0.93 (12H, d, J=5Hz)	2x -CH- CH ₃ CH ₃

Tab.1: ¹H-nmr data from 1 and tetrahydro-1 (CDCl₃-solution; 90 MHz, standard: TMS)

The chemical shifts of the non-indolic protons of 1 are in agreement with the presence of two γ,γ-dimethylallylic groups.

Irradiation experiments upon 1 and the observed alteration of the ¹H-nmr spectrum after hydrogenation prove fully a bis(γ,γ-dimethylallyl)-substituted indole structure.

The position of one of the substituents of the indole system at C-6 could readily be deduced from the ¹H-nmr data^{1,2}): In the case of a substituent at C-5 the characteristic signal of H-(4) at δ 7.48 ppm should appear as a doublet with J_{4,6}=1.6 Hz³).

From the singlet at δ 7.10 ppm the other substituent was supposed to be at C-3⁴). This was confirmed by the ¹³C-nmr data: According to Parker and Roberts alkylation of indoles at position-2 or -3 causes a characteristic downfield shift (10-13 ppm) in the ¹³C-resonance of the substituted carbon atom⁵). The values of interest from indole and the methyl indoles are listed in Table 2,

which gives, in addition, the corresponding signals found in the spectrum of 1

atom	indole	2-methylindole	3-methylindole	<u>1</u>
C-2	125.2 ⁵⁾ 123.7 ⁶⁾	135.7 ⁵⁾	122.7 ⁵⁾	123.3
C-3	102.6 ⁵⁾ 101.8 ⁶⁾	100.4 ⁵⁾	111.4 ⁵⁾	110.3

Tab.2: ¹³C-resonance of C-2 and C-3 in 1 and some indoles (ppm in CDCl₃-solution, standard: TMS).

Since there are no other signals between δ 77 ppm ($=^{13}\text{CDCl}_3$) and δ 110.3 ppm in the ¹³C-nmr spectrum of 1 carbon C-3 must be substituted.

Whereas a very few mono-substituted isoprenyl- or γ,γ -dimethylallyl-indoles are known to occur in plants^{1,7)} 3,6-bis(γ,γ -dimethylallyl)indole (1), isolated from *Uvaria elliotiana*, is the first naturally occurring indole substituted only by two unlinked isoprenoid units.

Furthermore it has to be pointed out, that 1 has been isolated as the main alkaloid from a species of the family Annonaceae, whose plants preferably produce alkaloids of the benzyl-isoquinoline-aporphine group^{8,9)}.

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